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HPAS (Main)-2012

PHYSICS

Paper II

Time : 3 Hours

Maximum Marks: 150

Note:— Question No. 1 is compulsory. Attempt any four questions from the rest. All questions carry equal marks.

1. Answer the following :

- (a) A small quantity of arsenic is doped in pure germanium crystal. How is the number of electrons and holes affected?
- (b) When does the avalanche breakdown occur in a p-n junction? Name a device based on it. 6
- (c) Two big metallic plates of area 1 m² each is placed

 1 m apart. The charge on the surface of the plates
 are equal and opposite. If the electric field between
 two plates is 5 N/C, then find the charge on each
 plate.

- (d) A condenser of capacity 0.5 μF is charged through a resistance of 10 M Ω . How much time will it take to discharge to half of its charge ? $(\log_e 2 = 0.6931)$
- (e) What is meant by the depth of modulation? Explain the meaning of cent-percent modulation. 6
- 2. (a) A gas has two specific heats, why? Differentiate them and explain why is the molar specific heat of gas at constant pressure C_p greater than the molar specific heat at constant volume C_V. 10
 - (b) Use law of equipartition of energy to obtain an expression for the molar specific heat of a diatomic gas at constant volume. How does the theoretical value differ from the experimental value?

 Explain.

- (c) Explain the quantum theory of rotational motion of a diatomic gas and find the contribution to specific heat.
- 3. (a) An isolated atom has its discrete energy levels.
 But when these atoms form a crystal, the energy levels form a band. Explain why?
 - (b) In the common base mode of a NPN transistor, the input resistance is 50 Ω and current gain is 0.96. If the load resistance in the output circuit is 5 kΩ, calculate its voltage and power gains.
 - (c) What is the Bark-Hausen's criterion for an oscillator? In which mode and how is transistor used for an oscillator?

- (a) Obtain an expression for the energy of an electron and wave number in a mono-valent atom according to the Bohr's model.
 - (b) Discuss the different series obtained in the spectrum of hydrogen atom on the basis of Bohr's model. What are the facts which could not be explained by this model?
 - (c) On the basis of the concept of de-Broglie waves, explain the stationary electron orbits of Bohr's atomic model.

- (b) Discuss the interpretation of matter wave function
 as given by Schrödinger and Max Born. How do
 the two concepts differ? Which concept is more
 correct and why?
- (c) What is meant by normalisation of a wave function? Obtain the condition of normalisation of a wave function.
- (a) Give the probabilistic interpretation of the wave function Ψ.
 - (b) Explain the meaning of particle in a box with the help of potential diagram. If for the position $0 \le x \le a$, one-dimensional potential field is $V_x = 0$, obtain the solution of Schrödinger equation. Normalise the wave function.

- (c) Calculate the transmission coefficient of a particle through a potential step of height V₀, if the energy of the particle is E when :
 - $(i) \cdot E > V_0$
 - (ii) E < V₀.
- (a) Show that the value of electronic polarization in vacuum is zero.
 - (b) Explain the concept of electrical susceptibility with the help of a capacitor filled with a dielectric medium and establish the relationship between dielectric constant and electrical susceptibility.
 - (c) What do you understand by sharpness of resonance in series resonant circuit? Find the expression for Q-factor of the circuit.

- (b) Distinguish between the nuclear fission and nuclear fusion with examples. By which of the two processes energy is obtained in :
 - (i) nuclear reactor,
 - (ii) in stars ?

Why is fusion reaction comparatively more superior?

(c) A nuclear reactor produces 30 mega watt electrical energy. If 200 MeV energy is released per fission,

Physics-II

Calculate:

- (i) The number of U²³⁵ nuclei fissioning per second,
- (ii) The mass of U²³⁵ required to produce electricity for 1000 hours.

(Avogadro number = 6.02×10^{23} per gm atom).